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© 2001 International Union of Crystallography Printed in Denmark – all rights reserved Cubic F432 crystals of recombinant mouse L-chain apoferritin were obtained by the hanging-drop technique with ammonium sulfate and cadmium sulfate as precipitants. The structure was refined to 2.1 and 1.6 Å resolution from data obtained at room temperature and under cryogenic conditions, respectively. The structure of an eight-amino-acid loop insertion in the mouse sequence is found to be highly disordered both at room temperature and at low temperature.

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## 1. Introduction

Ferritin, the ubiquitous iron-storage protein, has been largely investigated, but unravelling the different features of the structure-function relationships in this protein is still a matter of concern (for reviews, see Harrison & Arosio, 1996; Harrison *et al.*, 1998; Chasteen & Harrison, 1999). Its three-dimensional multimeric structure, which was determined 20 y ago (Clegg *et al.*, 1980), and its heteropolymeric composition in vertebrates (Adelman *et al.*, 1975) as well as the multiple active sites (iron entry-route channels, ferroxidase centre, mineral nucleation site) are some of the major factors which account for the complexity of its function.

Ferritins are multimeric proteins consisting of 24 subunits. Each of them (15–20 kDa) consists of a four  $(a, b, c, d) \alpha$ -helix bundle to which a fifth small  $\alpha$ -helix e is attached at the C-terminal end. The 24 subunits assemble in a 432 point symmetry. The multimer has a spherical shape (diameter 120–130 Å), with an inner cavity of about 80 Å diameter; this cavity can host up to 4500 iron ions stored as hydrous ferric oxide. Natural ferritins isolated from vertebrates are heteropolymers made of various types of subunits, named H, L or M, with different sequences, whereas bacterial and plant ferritins have been shown to be mainly homopolymers of H-type chains, the subunit associated with a ferroxidase activity.

Several structures of ferritin have been refined to a resolution of 2.0 Å or better: recombinant bullfrog L-chain ferritin (rBfLF; Trikha *et al.*, 1995; PDB code 1rcd), recombinant horse L-chain ferritin (rHoLF; Gallois *et al.*, 1997; PDB code 1dat; Hempstead *et al.*, 1997; PDB code 1aew), variant human H-chain ferritin (rHuHF; Hempstead *et al.*, 1997; PDB code 2fha), the best resolution (1.85 Å) being reached for the variant bullfrog H-chain ferritin (rBfHF; Takagi *et al.*, 1998; PDB code 1bg7).

In the course of our structural investigation on recombinant mouse L-chain ferritin (rMoLF; Granier *et al.*, 2000), we obtained three crystal forms obtained from different crystallization conditions: a first set of crystals (tetragonal and monoclinic forms) was obtained from solutions which did not contain any metal-ion salt and diffracted to a resolution of

#### Table 1

Data-collection statistics.

Values in	parentheses	correspond	to the	e last	resolution	shell

Temperature (K)	300	100
Crustel dimensions (mm)	$0.25 \times 0.22 \times 0.15$	$0.62 \times 0.60 \times 0.15$
Crystal dimensions (mm)	0.33 × 0.35 × 0.13	0.62 × 0.60 × 0.13
Crystal-to-film distance (mm)	159	95
Oscillation range $\varphi$ (°)	24.0	22.5
$\Delta \varphi$ (°) rotation per frame	1.5	0.9
Exposure time per frame (s)	900	4800
Unit-cell parameters (Å)	182.5	180.82
Resolution range (Å)	20-2.10	20-1.60
Last shell (Å)	2.21-2.10	1.64-1.60
N <sub>measured</sub>	83602 (10310)	158991 (8894)
Nunique	15681 (2030)	32292 (2042)
Completeness (%)	98.5 (91.3)	95.7 (83.8)
Multiplicity	5.6 (5.3)	4.9 (4.4)
$R_{\rm meas}$ †	0.141 (0.48)	0.079 (0.43)
$R_{\rm sym}$ †	0.126 (0.40)	0.071 (0.38)
PCV†	0.150 (0.56)	0.094 (0.56)
$\langle I/\sigma(I) \rangle$	6.0 (1.9)	9.3 (1.9)
$[I > 3\sigma(I)] (\%)$	87 (75)	79 (54)

† See Diederichs & Karplus (1997) for definitions.

2.46 Å. Using cadmium sulfate as the crystallization agent, we obtained crystals of the cubic F432 form isomorphous to rHoLF, variant rHuHF, rBfLF and rBfHF crystals. Cubic crystals of rMoLF diffract to 2.1 Å at room temperature. Under cryogenic conditions and using Cu Ka X-ray radiation generated by a rotating anode, we collected data to a resolution of 1.6 Å. Among the reasons which led us to investigate the structure of rMoLF, one of them concerns the specific mouse L-chain sequence: it exhibits, along with rat L-chain ferritin sequence, an eight-amino-acid insertion PAQTGAPQ in the exposed loop connecting helices d and e. In a previous study of recombinant rat L-chain ferritin crystals (Lawson, 1990), this insertion was found to be highly disordered. We present here the structure of the cubic form of rMoLF refined to a resolution of 1.6 Å. We briefly compare it with the roomtemperature structure refined to 2.1 Å resolution: in both cases, the de loop insertion is highly disordered. Nevertheless, the low-temperature structure allowed us to improve the model: a few more partially occupied cadmium-binding sites and their coordination shell pattern were determined, several additional side chains were positioned and the ordered water shell was notably built up. Differences from rHoLF and rBfLF are also briefly described.

# 2. Crystallization and data collection

Expression and purification have been described previously (Santambrogio *et al.*, 2000). An Ala residue replaces the reported Thr at position 121 (Beaumont *et al.*, 1989), a substitution arising from a  $G \rightarrow A$  transition possibly caused by a PCR error. The substitution did not affect protein assembly, stability or ferritin iron-uptake functionality (not shown). The subunit molecular weight is 20 641 Da (182 amino acids). Crystallization was performed using the hanging-drop vapour-diffusion method in Linbro plates. The best crystals were obtained in a few days at 293 K in hanging drops produced by

# Table 2

Refinement parameters.

Values in parentheses correspond to the last resolution shell.

Temperature (K)		300		100	
Resolution shell (	Å)	20.0-2.10	)	19.0-1.60	
Last shell (Å)	<i>,</i>	2.21-2.10	)	1.64-1.60	
R <sub>fac</sub>		0.16 (0.1)	5)	0.16 (0.22)	
R <sub>free</sub>		0.20 (016	5)	0.19 (0.27)	
DPI† (Å)		0.143	,	0.071	
EME‡ (Å)		0.07		0.05	
R.m.s. deviations					
Bonds (Å)		0.009		0.009	
Angle distances (A	Å)	0.023		0.022	
Planar 1-4 distanc	es (Å)	0.027		0.027	
	300 K		100 K		
Protein atoms	No.	$\langle B \rangle$ (r.m.s.) (Å <sup>2</sup> )	No.	$\langle B \rangle$ (r.m.s.) (Å <sup>2</sup> )	
Main chain	672	13.6 (4.5)	677	8.41 (2.9)	
Side chain	653	17.6 (7.3)	694	11.5 (6.1)	
Water molecules					
First shell	124	33.0 (12.8)	192	22.9 (10.8)	
Second shell	27	40.6 (8.5)	74	32.0 (11.7)	
Third shell	_	_ ` ´	20	38.6 (8.8)	
Cd <sup>2+</sup> ions	4	32.8 (10.3)	9	20.8 (5.8)	

† Diffraction-component precision index as defined by Cruickshank (1999). ‡ Expected maximal error. See definition in Vaguine *et al.* (1999).

mixing 3 µl of protein solution at a concentration of 3.5 mg ml<sup>-1</sup> in 20 mM Tris pH 7.4 and 3 µl of precipitant solution composed of 0.92 M ammonium sulfate, 0.4% CdSO<sub>4</sub> and 3 mM NaN<sub>3</sub>. Drops were equilibrated against 1 ml of the same precipitant solution. Crystals had approximate dimensions of  $0.5 \times 0.5 \times 0.15$  mm. X-ray diffraction experiments were conducted using an X-ray source (graphite monochromated Cu K $\alpha$ ,  $\lambda = 1.5418$  Å) provided by an Enraf-Nonius FR571 rotating-anode generator operating at 40 kV and 50 mA. Diffraction data were collected on a 300 mm MAR Research image-plate scanner, processed with the program MOSFLM (Leslie et al., 1986) and further scaled with the program SCALA (Collaborative Computational Project, Number 4, 1994). The data statistics are gathered in Table 1. Low-temperature data were obtained by the flash-freezing technique. 30% glycerol was added to the mother liquor. Low temperature (100 K) was obtained by cold nitrogen-gas flow using the Cryostream cooler (Oxford Cryosystems, England). Crystals were mounted on a modified Huber goniometer head equipped with a removable arc (Litt et al., 1998).

# 3. Refinement

The room-temperature and low-temperature structures were refined using the program *REFMAC* (Murshudov *et al.*, 1997). The starting model was that of rHoLF (Gallois *et al.*, 1997; atomic coordinates from PDB file 1dat). Models were visualized and modified based on  $2F_o - F_c$  and  $F_o - F_c$  electrondensity maps using the graphics programs *TURBO-FRODO* (Roussel *et al.*, 1990) and *XtalView* (McRee, 1993). Water molecules were positioned in well defined  $F_o - F_c$  residual

# Table 3 Intrasubunit and intersubunit hydrogen bonds and salt bridges observed in rMoL F and rHoL F

Side chain-main chainMain chain-side chainSide chain-side chainSide chain-side chainDonorAcceptorDonorAcceptorDonorAcceptorHydrogen bonds and salt bridges within a subunitArg727Gln60Xarg727Gln69Asn17Arg727Gln61Val121Ser9Gln69Gln13Tyr301Gln103Gln697Glu13Gly461Glu175Tp897Ser32Asn17Arg72Glu171Thr927Glu171Asn21Asp80Lcu1251Asp122Gln93Glu93Glu97Glu171Asp122Gln867Glu88Ser112Ser371Asp123Asp123Asp123Asp123Asp123Asp133Asp133Asp133Asp133Asp144Arg1537Asp144Glu130Ser1371Asp144Glu130Ser1371Asp127His114Thr91 $\ddagger$ Asp94Hys133Glu13Arg230Glu143Arg230Glu144Arg230Glu130Ser157Asp144Arg130Asp174Ser157Asp144Arg230Glu13Arg230Glu33Asp144Arg230Glu33Asp144Arg230Glu34Asp144Arg153Asp144Arg1537Asp144Arg1537Asp144Arg1537Asp145Arg1537Asp145Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Arg1537Asp145Asp145Arg1537Asp145 </th <th>u</th> <th>na moder.</th> <th></th> <th></th> <th></th> <th></th>	u	na moder.						
$ \begin{array}{ c c c c c c } \hline Donor & Acceptor & Donor & Acceptor & Donor & Acceptor \\ \hline Hydrogen bonds and salt bridges within a subunit \\ Arg5† Tyr8 & IIe4 & Ser2 & Tyr8† Gln69 \\ Arg72† Gln6 & Val12† Ser9 & Gln13 & Tyr30† & Glu103 \\ Gln69† Glu13 & Gly46† & Glu175 & Trp89† Ser32 \\ Asn17 & Arg75 & Lys91† & Glu04 & Arg72 & His124 \\ His23 & Leu19 & Thr92† & Glu171 & Thr92† & Glu171 \\ Asn21 & Asp80 & Leu125† & Asp122 & Gln93 & Glu97 \\ Thr29† & Arg25 & Glu171 & Ser168 & Gln69* & Asp112 \\ Ser27† & Ala55 & Lys142 & Asn150 \\ Gln86† & Glu88 & & & & & \\ Ser118† & His114 & & Thr91‡ & Asp94 \\ Cys126† & Asp122 & & His124 & & & & \\ Ser118† & His114 & & Thr91‡ & Asp94 \\ Cys126† & Asp122 & & & & & \\ Ser118† & His12 & & & & & \\ Thr49† & Gly145 & & & & & & \\ Asn150f & Asn146 & & & & & & \\ Cys48‡ & Gly34 & & & & & & \\ Tyr36$ & Gly90 & & & & & & & \\ Arg75$ & Glu71 & & & & & & & \\ Arg75$ & Glu13 & & & & & & & \\ Arg75$ & Glu13 & & & & & & \\ Arg75$ & Glu13 & & & & & & \\ Arg239 & Glu88 & & & & & & & \\ Lys58* & Glu130 & & & & & & & \\ Arg39 & Glu88 & & & & & & & \\ Lys58* & Glu137 & & & & & & & \\ Arg39 & Glu88 & & & & & & & \\ Lys58* & Glu137 & & & & & & & \\ Arg39 & Glu88 & & & & & & & \\ Lys58* & Glu137 & & & & & & & \\ Arg39 & Glu88 & & & & & & & \\ Lys58* & Glu137 & & & & & & & \\ Arg120 & Glu11 & & & & & & & \\ Arg39 & Glu88 & & & & & & & \\ Lys58* & Glu137 & & & & & & & \\ Arg170† & Glu175 & & & & & & & & \\ Lys64* & Glu79 & Lys83† & Asp80 & Ser2† & Asp40 \\ Lys67 & Asp38 & & & & & & \\ Hydrogen bonds and slt bridges within a dimer \\ Tyr28* & Glu79 & Lys83† & Asp80 & Ser2† & Asp40 \\ Lys67 & Asp38 & & & & & \\ Hydrogen bonds within a trimer \\ Lys104 & Gln3 & & & & & & \\ Arg152 & Gln3 & & & & & \\ Gln34 & Gly145 & & & & & & \\ His124 & Asp135 & & & & \\ Gln34 & Gly145 & & & & & \\ Hydrogen bonds within a trimer \\ Lys104 & Arg5 & & & & & & \\ Arg153 & Val42 & & & & & \\ Arg153 & Val42 & & & & \\ Arg153 & Val42 & & & & \\ Arg153 & Val42 & & & \\ Arg153 & Val42 & & & & \\ Arg163 & Leu161$ & &$	Side chain-r	nain chain	Main chain-	-side chain	Side chain-	side chain		
Hydrogen bonds and salt bridges within a subunit         Arg $25^{\uparrow}$ Tyr8       ILe4       Ser2       Tyr8       Gln69         Arg $72^{\uparrow}$ Gln6       Val12 $^{\uparrow}$ Ser9       Gln69       Asn17         Asn7       Ala121       Arg 75       Glu13       Tyr30 $^{\uparrow}$ Glu103         Gln69       Glu13       Gly46 $^{\uparrow}$ Glu17       Trp89 $^{\uparrow}$ Ser32         Asn17       Arg 75       Lys91 $^{\uparrow}$ Glu171       Trp92 $^{\uparrow}$ Glu171         Asn21       Asp80       Leu125 $^{\circ}$ Asp122       Gln33       Glu97         Thr29 $^{\uparrow}$ Arg 25       Glu171       Ser163       Lys142       Asn164         Tyr133       Gly61       Ser164       Asp122       His114       Glu130         Ser1187       His14       Thr91 $^{\circ}$ Asp94       Sys126       Asp140         Sys136       His132       Thr149 $^{\circ}$ Glu130       Sys137       Asp150         Ser1147       Asp127       Lys142       Asp140       Asp140       Ser131 $^{\circ}$ Asp140         Arg 753       Glu74       Ser157 $^{\circ}$ Arg153       Satt bridges       Satt bridges       Satt pridge       Satt pridge       Sa	Donor	Acceptor	Donor	Acceptor	Donor	Acceptor		
$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	Hydrogen b	onds and salt l	bridges within	n a subunit				
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Aro5t	Tvr8	Ile4	Ser?	Tyr8†	Gln69		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Arg72+	Gln6	Val12 <sup>+</sup>	Ser0	Gln69	Asn17		
Image of the second	Asn7†	Ala121	Arg75†	Glu13	Tyr30†	Glu103		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Gln69 <sup>+</sup>	Glu13	Glv46†	Glu175	Trp89†	Ser32		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Asn17	Arg75	Lys91†	Glu94	Arg72	His124		
Latz       Leu12st       Asp122       Glu193       Glu197         Thr29t       Arg25       Glu171       Ser168       Gln108†       Asp112         Ser27t       Ala55       Lys142       Asn146       Arg153†       Asn150         Gln86†       Glu88       Arg153†       Asn150       Glu171       Ser118†       His114       Thr91‡       Asp94         Cys126†       Asp122       His114‡       Glu130       Ser111†       Asp14       Glu130         Ser115†       Asp127       Lys136       His132       His114‡       Glu130       Ser105‡       Val101       Ser105‡       Val101       Ser105‡       Val101       Ser105‡       Val101       Ser105‡       Val101       Ser105‡       Val101       Ser17‡       Arg153       Satl bridges       Arg120       Glu11       Arg39†       Asp41       Arg120       Glu111       Arg39†       Asp41       Arg120       Alg175       Arg174       Ser157       Arg152       Glu97       Arg174       Ser17*       Asp40       Arg174       Asp40       Arg174       Asp40       Asp38       Asp40       Asp38       Asp40       Asp38       Asp40       Asp38       Asp40       Asp38       Asp40       Asp38       Asp38       Asp40 <td>His23</td> <td>Leu19</td> <td>Thr92†</td> <td>Glu171</td> <td>Thr92†</td> <td>Glu171</td>	His23	Leu19	Thr92†	Glu171	Thr92†	Glu171		
	Asn21	Asp80	Leu125†	Asn122	Gln93	Glu97		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Thr29*	Arg25	Glu171	Ser168	Gln108†	Asp112		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Ser27†	Ala55	Glui/I	501100	Lys142	Asn146		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Tyr133	Glv61			Arg153+	Asn150		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Gln86†	Glu88			/Hg155	71311130		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Ser118 <sup>+</sup>	His114			Thr91+	Asp94		
	Cvs126 <sup>+</sup>	Asp122			His114+	Glu130		
	Ser131+	Asp122			11131144	Gluiso		
	L vs136	His132						
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Thr149†	Glv145						
$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	Asn150	Asn146						
	Cvs48†	Glv34						
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Tvr36†	Gly90						
Arg 50+       Val 101         Ser105‡       Val 101         Ser157‡       Arg 153         Salt bridges       Arg 51+         Arg 20       Glu 13         Arg 39+       Asp 41         Arg 39+       Glu 88         Lys 58+       Glu 103         Lys 58+       Glu 103         Lys 58+       Glu 175         Lys 142‡       Asp 126         Arg 176+       Glu 75         Lys 142‡       Asp 146         Hydrogen bonds and salt bridges within a dimer         Tyr 28+       Glu 79         Lys 83+       Asp 80         Ser 22+       Asp 40         Asn 70‡       Asp 38         Val 81 \$       Salt bridges within a dimer         Arg 75       Asp 40         Lys 67       Asp 38         Hydrogen bonds within a trimer       Lys 104         Lys 104+       Gln 3       Asn 107 †         Lys 104       Arg 5       Arg 152         Gln 3†       Gly 145       Salt bridges within a trimer         Lys 104       Arg 5       Arg 152         Gln 3†       Gly 145       Salt bridges within a tetramer         Arg 152       Lys 142       Asp	Aro75†	Gly74						
Scriber information in the second se	Ser105†	Val101						
Salt bridges Arg5 $\ddagger$ Glu13 Arg120 Glu11 Arg39 $\ddagger$ Asp41 Arg39 Glu88 Lys58 $\ddagger$ Glu103 Lys58 $\ddagger$ Glu103 Lys58 $\ddagger$ Glu107 Arg72 $\ddagger$ Asp122 Arg152 Glu97 Arg176 $\ddagger$ Glu75 Lys142 $\ddagger$ Asp146 Hydrogen bonds and salt bridges within a dimer Tyr28 $\ddagger$ Glu79 Lys83 $\ddagger$ Asp80 Ser2 $\ddagger$ Asp40 Asn70 $\ddagger$ Phe35 Gln3 $\ddagger$ Asp40 Asn70 $\ddagger$ Phe35 Gln3 $\ddagger$ Asp40 Asn70 $\ddagger$ Asp38 Val81 $\ddagger$ Val81 $\$$ Salt bridges within a dimer Arg75 Asp40 Lys67 Asp38 Hydrogen bonds within a trimer Lys104 $\ddagger$ Gln3 Asn107 $\ddagger$ Gln6 Lys104 Arg5 Arg152 Gln3 Gln6 $\ddagger$ Lys104 His124 Asp135 Gln3 $\ddagger$ Gly145 Salt bridges within a tetramer Arg176 $\ddagger$ Leu177 Tyr172 $\ddagger$ Thr178 Lys139 Asp71 Hydrogen bonds within a tetramer Arg176 $\ddagger$ Leu177 Tyr172 $\ddagger$ Thr178 Lys142 $\ddagger$ Asp38	Ser157†	Arg153						
Sart Site       Glu13         Arg5 <sup>†</sup> Glu13         Arg39 <sup>†</sup> Asp41         Arg39       Glu88         Lys58 <sup>†</sup> Glu103         Lys58 <sup>†</sup> Glu170         Arg176 <sup>†</sup> Glu77         Arg176 <sup>†</sup> Glu77         Arg176 <sup>†</sup> Glu79         Lys88 <sup>†</sup> Asp146         Hydrogen bonds and salt bridges within a dimer         Tyr28 <sup>†</sup> Glu79         Lys83 <sup>†</sup> Asp80         Ser2 <sup>†</sup> Asp40         Asn70 <sup>‡</sup> Phe35         Glu3 <sup>†</sup> Asp40         Asn70 <sup>‡</sup> Phe35         Salt bridges within a dimer       Asn70 <sup>‡</sup> Arg75       Asp40         Lys67       Asp38         Hydrogen bonds within a trimer       Lys104         Lys104 <sup>†</sup> Gln3         Gln3 <sup>†</sup> Gly145         Salt bridges within a trimer       Lys139         Lys139       Asp71         Hydrogen bonds within a tetramer       Arg152         Gln3 <sup>†</sup> Gly145         Salt bridges within a tetramer       Arg176 <sup>†</sup> Lys139       Asp71         Hydrogen bonds within a tetramer<	Salt bridges	riigiss						
Arg120       Glu11         Arg39†       Asp41         Arg39       Glu88         Lys58†       Glu103         Lys58†       Glu173         Arg122       Glu79         Arg176†       Glu175         Lys142‡       Asp146         Hydrogen bonds and salt bridges within a dimer         Tyr28†       Glu79         Lys83†       Asp80         Ser2†       Asp40         Asn70†       Phe35         Gln3†       Asp40         Asn70‡       Asp38         Val81‡       Val81§         Salt bridges within a dimer       Asr70‡         Arg75       Asp40         Lys67       Asp38         Val81‡       Val81§         Salt bridges within a dimer       Arg152         Arg75       Asp40         Lys67       Asp38         Hydrogen bonds within a trimer       Lys104         Lys104       Arg5       Arg152         Gln3†       Gly145         Salt bridges within a tetramer       Arg156†         Arg156†       Leu177       Tyr172†         Hydrogen bonds within a tetramer       Arg176†         Arg156†       Leu1	Arø5†	Glu13						
Arg39       Asp41         Arg39       Glu88         Lys58†       Glu103         Lys58†       Glu137         Arg72†       Asp122         Arg152       Glu97         Arg176†       Glu175         Lys142‡       Asp146         Hydrogen bonds and salt bridges within a dimer         Tyr28†       Glu79         Lys83†       Asp80         Ser2†       Asp40         Asn70†       Phe35         Gln3†       Asp40         Asr70‡       Asp38         Val81‡       Val81§         Salt bridges within a dimer       Arg152         Arg75       Asp40         Lys67       Asp38         Hydrogen bonds within a trimer       Lys104†         Lys104†       Gln3       Asn107†         Lys104       Arg5       Arg152       Gln3         Gln6†       Lys104       His124       Asp135         Gln3†       Gly145       Salt bridges within a trimer       Lys139       Asp71         Hydrogen bonds within a tetramer       Arg176†       Leu177       Tyr172†       Thr178         Lys142†       Asp38       Thr149†       Asp40       Arg153†	Arg120	Glu11						
Arg39Glu88Lys58†Glu103Lys58†Glu137Arg72†Asp122Arg152Glu97Arg176†Glu175Lys142‡Asp146Hydrogen bonds and salt bridges within a dimerTyr28†Glu79Lys83†Asp80Ser2†Asp40Asn70†Phe35Gln3†Asp40Asn70‡Asp38Val81†§Val81§Salt bridges within a dimerArg75Asp40Lys07Asp38Hydrogen bonds within a trimerLys104†Gln3Lys104‡Gln3Gln6†Lys104Gli3†Gly145Salt bridges within a trimerLys104Arg5Gln3†Gly145Salt bridges within a tetramerArg176†Leu177Tyr172†Thr178Lys139Asp40Arg151†Val42Asn150†Ala43Arg168‡Leu169Lys142Ser157	Arg39†	Asp41						
Lys58†       Glu103         Lys58†       Glu137         Arg72†       Asp122         Arg152       Glu97         Arg176†       Glu175         Lys28‡       Asp146         Hydrogen bonds and salt bridges within a dimer         Tyr28†       Glu79         Lys83†       Asp80         Ser2†       Asp40         Asn70†       Phe35         Salt bridges within a dimer         Arg75       Asp40         Lys67       Asp38         Hydrogen bonds within a trimer         Lys104       Arg5         Gln3†       Asp107†         Gln61       Lys104         Hydrogen bonds within a trimer         Lys104       Arg5         Gln3†       Gly145         Salt bridges within a trimer         Lys104       Arg5         Gln3†       Gly145         Salt bridges within a trimer         Lys139       Asp71         Hydrogen bonds within a tetramer         Arg176†       Leu177         Lys142†       Asp38         Thr149†       Asp40         Arg153†       Val42         Asn150†       Ala43	Arg39	Glu88						
Lys58† Glu137 Arg72† Asp122 Arg152 Glu97 Arg176† Glu175 Lys142‡ Asp146 Hydrogen bonds and salt bridges within a dimer Tyr28† Glu79 Lys83† Asp80 Ser2† Asp40 Asn70† Phe35 Gln3† Asp40 Asn70‡ Asp38 Val81†§ Val81§ Salt bridges within a dimer Arg75 Asp40 Lys104† Gln3 Asn107† Gln6 Lys104† Gln3 Asn107† Gln6 Lys104 Arg5 Arg152 Gln3 Gln6† Lys104 His124 Asp135 Gln3† Gly145 Salt bridges within a trimer Lys104 Arg5 Asp10 His124 Asp135 Gln3† Gly145 Salt bridges within a tetramer Arg176† Leu177 Tyr172† Thr178 Lys142† Asp38 Thr149† Asp40 Arg153† Val42 Asn150† Ala43 Arg168‡ Leu169 Leu161‡ Ser157	Lys58†	Glu103						
Arg72†Asp122Arg152Glu97Arg176†Glu175Lys142‡Asp146Hydrogen bonds and salt bridges within a dimerTyr28†Glu79Lys83†Asp80Ser2†Asp40Asn70†Phe35Glu3†Asp40Asn70‡Asp38Val81‡§Val81§Salt bridges within a dimerArg75Asp40Lys107Asp38Hydrogen bonds within a trimerLys104Gln3Lys104Arg5Gln3†Gly145Salt bridges within a trimerLys104Arg5Gln3†Gly145Salt bridges within a trimerLys139Asp71Hydrogen bonds within a tetramerArg176†Leu177Tyr172†Thr178Lys142†Asp38Thr149†Asp40Arg153†Val42Asn150†Ala43Arg168‡Leu161‡Ser157	Lvs58†	Glu137						
Arg152Glu97Arg176†Glu175Lys142‡Asp146Hydrogen bonds and salt bridges within a dimerTyr28†Glu79Lys83†Asp80Ser2†Asp40Asn70†Phe35Gln3†Asp40Asn70‡Asp38Val81†§Val81§Salt bridges within a dimerArg75Asp40Lys67Asp38Hydrogen bonds within a trimerLys104Gln3Lys104Arg5Gln6†Lys104His124Asp135Gln3†Gly145Salt bridges within a trimerLys139Asp71Hydrogen bonds within a tetramerArg16†Leu177Tyr172†Thr178Lys142†Asp38Thr149†Asp40Arg153†Val42Asn150†Ala43Arg168‡Leu169Leu161‡Ser157	Arg72†	Asp122						
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Arg152	Glu97						
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$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	Asn70†	Phe35			Gln3†	Asp40		
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$\begin{array}{llllllllllllllllllllllllllllllllllll$	Hydrogen be	onds within a	trimer					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Lys104†	Gln3			Asn107†	Gln6		
	Lys104	Arg5			Arg152	Gln3		
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Salt bridges within a trimerLys139Asp71Hydrogen bonds within a tetramerArg176†Leu177Tyr172†Thr178Lys142†Asp38Thr149†Asp40Arg153†Val42Asn150†Ala43Arg168‡Leu169Leu161‡Ser157	Gln3†	Gly145						
Lys139Asp71Hydrogen bonds within a tetramerArg176†Leu177Tyr172†Thr178Lys142†Asp38Thr149†Asp40Arg153†Val42Asn150†Ala43Arg168‡Leu169Leu161‡Ser157	Salt bridges	within a trime	er					
Hydrogen bonds within a tetramer Arg176† Leu177 Tyr172† Thr178 Lys142† Asp38 Thr149† Asp40 Arg153† Val42 Asn150† Ala43 Arg168‡ Leu169 Leu161‡ Ser157	Lys139	Asp71						
Arg176†         Leu177         Tyr172†         Thr178           Lys142†         Asp38         Thr149†         Asp40           Arg153†         Val42         Asn150†         Ala43           Arg168‡         Leu169         Leu161‡         Ser157	Hydrogen be	onds within a	tetramer					
Lys142†         Asp38           Thr149†         Asp40           Arg153†         Val42           Asn150†         Ala43           Arg168‡         Leu169         Leu161‡           Ser157	Arg176†	Leu177			Tyr172†	Thr178		
Thr149†         Asp40           Arg153†         Val42           Asn150†         Ala43           Arg168‡         Leu169         Leu161‡	Lys142†	Asp38						
Arg153†         Val42           Asn150†         Ala43           Arg168‡         Leu169         Leu161‡           Ser157	1hr149†	Asp40						
Asn150† Ala43 Arg168‡ Leu169 Leu161‡ Ser157	Arg153†	Val42						
Arg168‡ Leu169 Leu161‡ Ser157	Asn150†	Ala43		a				
	Arg168‡	Leu169	Leu161‡	Ser157				

 $\dagger$  Interactions in common with rHoLF.  $\ddagger$  Interactions observed in rHoLF and not in rMoLF. \$ Main chain-main chain.

densities with a lower cutoff of  $3\sigma$ . Cadmium ions were positioned according to anomalous Fourier difference maps with a lower cutoff of  $4\sigma$ . Refinement characteristics are gathered in Table 2.

#### 3.1. Model quality

 $2F_o - F_c$  maps contoured at  $1\sigma$  have continuous electron density along the subunit backbone, with the exception of the first N-terminus residue, Thr1, the 156-167 segment which corresponds to the de loop insertion and the two last C-terminal residues, i.e. His181 and Asp182. Of the remaining residues, several side chains could not be positioned, *i.e.* those of Glu11, Arg18, Arg25, Glu45, Glu53, Glu56, Lys91 and Arg120 for the room-temperature structure, and Arg25 and Lys91 for the low-temperature structure. Using the program SFCHECK (Vaguine et al., 1999), the lowest density correlation coefficients ( $D_{corr} = 0.85-0.88$ ) are obtained for the side chains of residues Asp38, Glu60, Glu67 and Gln79 for the room-temperature structure, and the side-chain atoms of residue Glu53 for the low-temperature structure. A Ramachandran plot (Ramachandran et al., 1963) shows that more than 95% of residues are positioned in the most favourable regions. All the residue side chains adopt standard conformations with the exception of two, Asp40 and Tyr28, the side chains of which adopt energetically slightly unfavourable  $\chi_1$ - $\chi_2$  torsion angles: the first residue, although being largely exposed to solvent, is involved in some electrostatic and hydrogen-bond interactions with water molecules, whereas for the second residue the unfavourable  $\chi_1 - \chi_2$  values arise from strong hydrophobic interactions and steric hindrance by neighbouring residues of the same subunit as well as by residues of the neighbouring subunit related by the twofoldsymmetry molecular axis.

#### 3.2. Structure of the subunit

The *ad*  $\alpha$ -helix bundle is defined by the segments Thr10– Phe37, Glu45–Arg72, Thr92–Ala119 and Pro123–Val154, whereas segment Leu169–Leu177 defines the fifth *e* helix located at the C-terminal extremity (Fig. 1*a*). Helices *b* and *c* are connected by a long loop Gly74–Gly90. Helix *d* has the characteristic bending at position His132 as found in rHoLF or rBfLF. The non-helical regions are characterized by the following.

(i) At the N-terminal end, two consecutive  $\beta$ -turns, *i.e.* residues Ser2–Arg5 (type I) and residues Arg5–Tyr8 (type IV).

(ii) Two successive type I (residues Arg39–Val42) and type IV (residues Asp40–Ala43)  $\beta$ -turns at the *ab* loop and a long twisted *bc* loop (residues Gly74–Gly90) which ends with a type VIII  $\beta$ -turn (residues Gln86–Trp89). The main-chain orientation of residue Val81 allows its NH and CO groups to build a short antiparallel pleated sheet with the same residue of the neighbouring subunit related by a twofold axis.

(iii) At the C-terminal end, the last Leu177–Lys180 residues with well defined electron density build a type IV  $\beta$ -turn.

#### 3.3. Comparison with room-temperature structure

Both protomer structures are very similar; the r.m.s. deviation of main-chain atoms is 0.15 Å. The only noticeable differences concern a few residue side chains which adopt different or alternate conformations: lowering the tempera-

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(a)



#### Figure 1

(a) Superposition of three subunits of rMoLF (green), rHoLF (white) and rBfLF (pink). (b) Stereoview of the N-terminal side of the *bc* loop fragment, for which a larger r.m.s. deviation on backbone atoms is observed between rMoLF (bold) and rHoLF (thin). (c) Stereoview of the C-terminal side of the *bc* loop in rMoLF (bold) and rBfLF (thin) structures.

ture allowed the positioning of six more side chains, 135 more water molecules and five more cadmium partial binding sites (see Table 2). The latter were determined, as mentioned above, with anomalous Fourier difference maps calculated with data in the range 8–2.0 Å.

#### 3.4. Comparison with rHoLF and rBfLF

This structure exhibits a few differences from those of 1dat and 1rcd, as revealed by the  $C^{\alpha}$  superposition of rHoLF and rBfLF onto those of rMoLF: r.m.s. deviations are 0.23 and 0.36 Å, respectively. For rHoLF, the most important differences occur in the *bc* loop and the poorly defined extremities of the *de* insertion loop of the rMoLF sequence. The deviations in the *bc* loop superposition is explained by the presence of a valine in position 81 instead of a leucine in rHoLF. The side-chain atoms of residue Val81 are embedded between the main chain of the *bc* loop and helix *a* and the bulkiness of Leu81 in rHoLF is responsible for the shift of residues 79–84 (Fig. 1*b*). For rBfLF, the greatest deviation occurs at the  $\pi$ -bulge position 132 in helix *d*, where a proline substitutes for

> a histidine in the bullfrog L-chain sequence. Differences in the backbone appear also at the end of the *bc* loop, where a type IV  $\beta$ -turn (Arg86–Trp89) is followed by a  $\gamma$ -turn (Tyr89–Asn91) (Fig. 1*c*) in the rBfLF structure.

#### 3.5. Intrasubunit hydrogen bonds and salt bridges

Polar and electrostatic interactions are listed in Table 3 and are compared with those of rHoLF, to which they are very similar (70% interactions conserved). A major difference with this latter concerns the salt bridge embedded within the fourhelix bundle: this salt bridge is a common feature of L-chain ferritins and substitutes the ferroxidase centre observed in H-chain ferritins. In the present structure, several differences from the horse L-chain sequence, i.e. Leu102Met, Phe133Tyr, Glu136Lys and particularly Tyr23His substitutions, noticeably modify the environment of the conserved Lys58-Glu103 electrostatic interaction: the Tyr23His substitution leads to the presence of a water molecule which binds His23  $N^{\varepsilon}$  and occupies the space held by Tyr23 OH in rHoLF: Fig. 2 depicts the penetration of water molecules inside the four-helix bundle. In rBfLF, an electrostatic interaction is observed between Lys23 and Glu58. These three examples show that a strong variability is allowed for this salt bridge, which has been mentioned as playing an important role in the thermal stability of L-chain ferritins (Hempstead et al., 1997; Martsev et al., 1998).

#### 3.6. Hydrophilic interactions upon shell assembly

Among the polar and salt-bridge interactions listed in Table 3, 24 involve residues from neigh-

bouring subunits. Interactions with solvent water molecules are depicted in more detail in the present structure, where 286 water molecules have been located, of which 192 belong to the first shell. Among these latter water molecules, 47 connect neighbouring subunits and are shown in Fig. 3 along with the intersubunit electrostatic and polar interactions. Fig. 3 underlines the small number of non-water interactions present in the vicinity of the threefold axes, the entry route of iron ions into the ferritin shell.

#### 3.7. Hydrophobic interactions upon shell assembly

The ferritin subunit exhibits on its external surface noticeable hydrophobic regions which are buried upon assembly of the protein shell. Two of these regions were previously depicted in the structures of rHoLF (Ford *et al.*, 1984) and of rHuHF (Lawson *et al.*, 1991; Hempstead *et al.*, 1997). An



#### Figure 2

The salt bridge embedded within the four  $\alpha$ -helix bundle of rMoLF (thick green bonds and ribbons, large-radius atoms) and rHoLF (thin yellow bonds and ribbons, small-radius atoms). Hydrogen bonds and electrostatic interactions are shown for the rMoLF structure only.



#### Figure 3

Stereoview of shell assembly. Subunit A is related to neighbours by 432 symmetry, subunit B by twofold symmetry, subunits C and E by fourfold symmetry, and subunit F and G by threefold symmetry. Main- and side-chain atoms of residues involved in intersubunit hydrogen bonds and salt-bridge interactions are shown, along with water molecules which bind atoms of two or more different subunits.

attempt to describe these hydrophobic surfaces is carried out using the software QUILT (Lijnzaad et al., 1996; Lijnzaad & Argos, 1997). The surface values of the six largest hydrophobic patches are listed in Table 4, along with those of their fractions which remain exposed to the solvent after the shell assembly. Patches 1 and 2 display large values (about 600  $Å^2$  each) and correspond to the interface regions generated by the twofold and fourfold symmetry axes: they are 80% buried when the protein shell is assembled. Less extended patches 3 and 4 are also 50 and 80% buried upon shell assembly. Fig. 4 shows the various proportions of accessible areas of one subunit, as well as their fractions buried upon shell assembly. It allows us to note the following. When subunits assemble, the proportion of buried hydrophobic and hydrophilic areas remains constant at about 60 and 40%, respectively. However, the contribution of the four largest hydrophobic patches to the buried hydrophobic area is predominant: whilst they represent 31% of the

accessible hydrophobic area (Fig. 4, column 1), their buried fraction upon shell assembly represents 56% of the hydrophobic buried area (Fig. 4, column 2). On the other hand, small hydrophobic patches, the areas of which are less than  $\sim 200 \text{ Å}^2$ , do not seem to contribute to shell assembly: the first water shell shields most of these patches as well as polar residues (Fig. 4, column 3). For example, patch number 6, listed in Table 4, is almost masked by ordered water molecules which bind neighbouring polar residues.

Therefore, the four largest hydrophobic patches listed in Table 4 definitely play an active role during shell assembly of the ferritin molecule.

#### 3.8. Metal-binding sites

Cadmium-binding sites were determined from anomalous Fourier difference maps. Residues involved in cadmium-binding sites are listed in Table 5. They are identical to those previously found in rHoLF (Granier et al., 1998 and references therein): Asp80 and Gln82 for intermolecular contacts; Glu53, Glu56, Glu57 and Glu60 for the ferrihydrite nucleation centre; and His114, Glu130, Asp127 and His132 for the threefold-axis entry route of iron ions. Only one exception, an additional site, is found at His49, close to the nucleation centre. Average distances are 2.19 and 2.27 Å for Cd-O and  $Cd-N^{\delta}$  (histidine) bonds, respectively. As regards the coordination of the various sites, only those of Asp80, Gln82 (square planar) and His132 (octahedral) are well defined, whereas those located on the threefold axis (His114, Glu130 and Asp127)

#### Table 4

Hydrophobic patches at the subunit external surface.

See Fig. 3 for labels for intersubunit contacts. Values are in  $\mathring{A}^2$ .

Patch No.	Residues involved	Symmetry involved	Accessible area for a single subunit	Remaining accessible area per subunit after shell assembly	After shell assembly including solvent
1	Tyr28, Phe35, Arg59, Glu63, Phe78, Gln79, Val81	Twofold axis A–B	603	100	51
2	Arg153, Ser168, Leu169, Tyr172, Leu173, Arg176	Fourfold axis A-C	591	116	59
3	Leu100, Ala101, Met102, Lys104, Ile141, Lys142, Glv145, Leu148	Threefold axis A-F	230	119	2
4	Leu111, His114, Glu130, Leu134	Threefold axis A-G	217	44	0
5	Leu22, Arg25, Gln82, Asn105, Ala109	Outer-shell side	175	175	45
6	Phe50, Leu54, Glu57, Lys136, Leu140, Lys143	Inner-shell side	176	176	7
Total	2		1992	730	164

#### Table 5

Residues involved in cadmium-binding sites.

Side-chain density correlation coefficients and average shifts (Å) are from SFCHECK.

	300 K		100 K		
Residue	D <sub>corr</sub>	Shift (Å)	$D_{\rm corr}$	Shift (Å)	
His49	0.95	0.17	0.96	0.13	
Glu53	_	_	0.90	0.13	
Glu56	_	_	0.96	0.09	
Glu57	0.91	0.16	0.95	0.10	
Asp80	1.0	0.04	1.0	0.06	
Gln82	0.98	0.10	0.94	0.09	
Glu60	0.89	0.19	0.98	0.05	
His114	1.0	0.06	0.99	0.02	
Cys126	1.0	0.04	0.97	0.03	
Asp127	1.0	0.04	1.0	0.07	
Glu130	0.98	0.09	1.0	0.07	
His132	0.99	0.07	1.0	0.03	

as well as those at the nucleation site (His49, Glu53, Glu56, Glu57, Glu60) are incompletely determined owing to threefold symmetry disorder, statistical disorder or incomplete determination of coordinated water molecules.

### 4. Conclusions

The structure of rMoLF has been determined at 1.6 Å resolution under cryogenic conditions. The protomer shows little differences from two other L-chain ferritin structures, *i.e.* those of horse and bullfrog. The structure of the eight-aminoacid insertion PAQTGAPQ, located in the *de* loop could not be determined at either room temperature or at low temperature. This part of the structure is strongly disordered as previously observed in the room-temperature structure of the recombinant rat L-chain ferritin (Lawson, 1990). The major difference from rHoLF and rBfLF concerns the salt bridge embedded within the fourhelix bundle of the subunit. In rMoLF, the presence of a histidine in position 23 instead of a tyrosine or a lysine is compensated by the penetration of water molecules into the helix bundle.

The hydrogen bonds and salt bridges are highly conserved: 24 of them participate in the shell assembly along with 47 water molecules. All interface regions are involved with the exception of the vicinity of the threefold-axis interface (the iron-entry route).

The metal-binding sites are identical to those observed in rHoLF, with the exception of a new site involving a histidine in position 49. This residue is highly conserved in mammalian L-chain ferritins and could be considered as part of the ferrihydrite nucleation centre.

Six large hydrophobic patches located at the subunit external surface have been described using the procedures of Lijnzaad *et al.* (1996). Four of them seem to actively participate in the assembly process of the ferritin subunits into the protein shell. The two largest patches correspond to interface



#### Figure 4

Variation of the hydrophilic (white) and hydrophobic (grey) surfaces ( $Å^2$ ) of a rMoLF subunit: accessible surface of a subunit (column 1), buried surface upon shell assembly (column 2) and non-accessible surface upon shell assembly, taking into account ordered water molecules of the first hydration shell (column 3). The contribution of the four largest hydrophobic patches is shaded in dark grey.

regions of subunits related by twofold and fourfold symmetries, whereas the third and fourth patches involve subunits related by threefold symmetry.

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